A New Approach to the Stereospecific Total Synthesis of Racemic Cecropia Juvenile Hormone[†]

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Summary 4-Methyl-5,6-dihydro-2*H*-thiopyran and its dimer (4) are used as structural units for the synthesis of C_{18} -juvenile hormone.

WE describe herein a new and economically feasible method for the stereospecific synthesis¹ of a racemic *Cecropia* juvenile hormone (C_{18} -J.H.), based on condensation of dihydrothiopyrans. The structure of C_{18} -J.H. can be divided into three structural units (A)—(C). The configuration of two ethyl groups in units (A) and (B) might be retained if they were blocked with two sulphur atoms [see (A') and (B')]. These blocking atoms could easily be removed by reductive desulphurization.

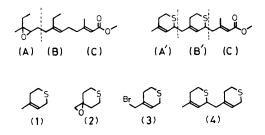
A key intermediate, the thiopyran (4), was synthesised by two methods. Condensation of tetrahydrothiopyran-4one² with dimethyloxosulphonium methylide afforded the epoxide (2)^{\ddagger} (65%), m.p. 52°. The carbanion prepared from the thiopyran (1)³ and BuⁿLi in the presence of 2,3-diazabicyclo[2,2,2]octane (DABCO) was treated with the epoxide (2) in THF at -20° and the resulting adduct was further dehydrated with SOCl₂-pyridine to give the desired dimeric dihydrothiopyran (4) [73% based on (2)],

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[‡] All new compounds gave satisfactory elemental analyses and i.r. and n.m.r. spectra consistent with the assigned structures.

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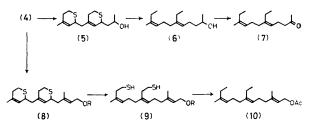
b.p. 85° at 0.2 mmHg. The other method was based on the condensation of (1) with (3), which was prepared from tetrahydrothiopyran-4-one by treatment with HCN, followed by solvolysis with EtOH, dehydration, reduction, and bromination with PBr₁.



Attachment of the final unit (C) and the removal of the blocking sulphur atoms were carried out by the following routes.

Route A: The dimer (4) was converted into a carbanion by treatment with BunLi-DABCO and was then treated with propylene oxide to afford the alcohol (5) (84%). Reductive desulphurization of (5) by treatment with metallic Li in ethylamine⁴ at -20° produced the dodecadienol (6) (60%), b.p. 97-98° at 0.45 mmHg. Oxidation of the alcohol (6) with Jones reagent yielded the ketone (7) (85%), b.p. $98-99^{\circ}$ at 1.5 mmHg, which was further transformed into deoxy C₁₈-J.H. by condensation with diethyl methoxycarbonylmethylphosphonate.5 Epoxidation⁶ of the ester gave racemic J.H., which was identical spectroscopically with natural J.H.7

Route B: The carbanion of dimer (4) was condensed with trans-4-chloro-3-methylbut-2-enyl tetrahydropyranyl ether, prepared from isoprene,⁸ to yield the triene (8; R = THP) (60%). The trienol (8; R = H) was treated with Li in ethylamine at -70° and the resulting dithio-alcohol (9; R = H) was converted into the corresponding acetate (9; R = Ac) (80%). Final desulphurization of (9; R = Ac) was achieved by treatment with excess of deactivated Raney nickel (W-2/acetone) to give the acetate (10) as a colourless oil (55%), b.p. 115° at 0.15 mmHg. Deoxy-C₁₈-J.H. obtained from (10) by the method of Corey et al.⁹ was identical spectroscopically with that prepared from the ketone (7).



 C_{18} -J.H. can thus be obtained with 100% stereospecificity at C-6 and C-10 and 95% or more at C-210 by route A, and with complete stereospecificity by route B.

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¹ E. J. Corey, J. A. Katzenellenbogen, N. W. Gilman, S. A. Roman, and B. W. Erickson, J. Amer. Chem. Soc., 19668, 90, 5618; R. Zurflüh, E. N. Wall, J. B. Siddall, and J. A. Edwards, *ibid.*, 6224; W. S. Johnson, T. Li, D. J. Faulkner, and S. F. Campbell, *ibid.*, p. 6225; E. J. Corey and H. Yamamoto, *ibid.*, 1970, 92, 6636; E. E. van Tamelen, P. McCurry, and U. Huber, Proc. Nat. Acad. Sci. U.S.A., 1971, 68, 1294.

- ² E. A. Fehnel and M. Carmack, J. Amer. Chem. Soc., 1948, 70, 1813.
- ³ R. F. Naylor, J. Chem. Soc., 1949, 2749.

- ³ R. F. Naylor, J. Chem. Soc., 1949, 2749.
 ⁴ A. S. Hallsworth, H. B. Henbest, and T. I. Wrigley, J. Chem. Soc., 1957, 1969.
 ⁵ K. H. Dahm, B. M. Trost, and H. Röller, J. Amer. Chem. Soc., 1967, 89, 5292.
 ⁶ E. E. van Tamelen and T. J. Curphey, Tetrahedron Letters, 1962, 121.
 ⁷ B. M. Trost, Accounts Chem. Res., 1970, 3, 120.
 ⁸ W. Oroshnik and R. A. Mallory, J. Amer. Chem. Soc., 1950, 72, 4608.
 ⁹ E. J. Corey, N. W. Gilman, and B. E. Ganem, J. Amer. Chem. Soc., 1968, 90, 5616.
 ⁹ V. Diverson, M. Murgi and T. Matsuwar, Nichon Kagagha, Zaski, 1969, 89, 431.
- ¹⁰ Y. Butsugan, M. Murai, and T. Matsuura, Nippon Kagaku Zasshi, 1969, 89, 431.